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Mn(III)-based radical addition reactions of 2-nitroindole with activated CH compounds

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ABSTRACT

2-Nitroindole undergoes addition reactions with the radicals generated from active CH compounds upon treatment with $Mn(OAc)_3 \cdot 2H_2O$ to afford the corresponding 3-substituted-2-nitroindoles in 27–66% yields. Products of the methylene addition reactions undergo a subsequent in situ Nef reaction to afford 2-oxoindolin-3-ylidenes.

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Over the past 40 years manganese(III)-promoted oxidative freeradical reactions have emerged as a powerful and versatile method for organic synthesis.¹ Recent applications include the tandem malonyl radical alkene addition and cyclization onto indole,² a catalytic-based addition of carboxyalkyl radicals to alkenes,³ and the key step in the first total synthesis of mersicarpine,⁴ to name but a few.

In continuation with our interest in the chemistry of 2- and 3nitroindoles,⁵ we now report that 2-nitroindole (1)⁶ undergoes manganese(III) acetate-promoted radical addition reactions with activated methylene and methine compounds in refluxing acetic acid (Scheme 1). Our results are summarized in Table 1. Whereas the reactions of **1** with 3-methylpentane-2,4-dione and 5-oxo-4propionylheptane-nitrile with Mn(OAc)₃·2H₂O (HOAc, reflux) afford the anticipated 3-substituted 2-nitroindoles **2a** and **2b**, respectively, the same reaction conditions with pentane-2,4-dione, nalonitrile, dimethyl malonate, 1,3-diphenylpropane-1,3-dione, 1-phenylbutane-1,3-dione, and methyl 3-oxobutanoate unexpectedly give the corresponding 2-oxoindolin-3-ylidenes (**4a-f**), the products of an in situ Nef reaction⁷ of intermediate **2**. Presumably, 2-nitroindole **2** undergoes tautomerization to a 2-*aci*-nitro species **3** followed by hydrolysis to **4**.

Our synthesis of **4** provides an alternative route to these 2-oxoindolin-3-ylidenes that typically involves condensation of isatin with activated methylene compounds.⁸ It might be noted that some 2-oxoindolin-3-ylidenes demonstrate selective inhibition of tyrosine kinases,⁹ and 3-methyleneindolin-2-one, which is a metabolite of indole-3-acetic acid, exhibits cytotoxicity and has potential for use in cancer therapy.¹⁰ We had previously found that 2-nitro-1-(phenylsulfonyl)indole reacts with enolates of diethyl malonate and cyclohexanone (anionic addition) to give 3-alkyl-2nitroindoles. However, those adducts are stable under the reaction conditions (NaH, THF, low temperature), and do not undergo a Nef reaction.¹¹

Interestingly, 3-nitroindole is unreactive under these conditions $(Mn(OAc)_3 \cdot 2H_2O, AcOH, reflux)$ with the same active CH compounds, which is perhaps due to the absence of a captodative stabilizing effect that the radical intermediate **A** generated from 2-nitroindole enjoys, whereas radical intermediate **B** does not (Fig. 1).

The structures of **2a–b** and **4a–f** are supported by spectral data (MS, NMR),¹² and the ¹H- and ¹³C-NMR data are consistent with literature values for similar compounds.^{13,14} Alkylidenes **4e,f**, which bear different substituents ($R^1 \neq R^2$), are obtained as mixtures of *E*- and *Z*-isomers. The stereochemistry at the C3–C8 double bond of isomeric **4e** could be deduced by comparison of the 1D-NOE



Scheme 1.

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Table 1

3-Alkyl-2-nitroindoles ${\bf 2a-b}$ and 2-oxoindolin-3-ylidenes ${\bf 4a-f}$ produced using $Mn(OAc)_3$ via Scheme 1^a

Entry	Compound	R ¹	R ²	R ³	Time (min)	Yield (%)
1	2a	CO(O)Me	CO(O)Me	Me	30	53
2	2b	C(O)Et	C(O)Et	$(CH_2)_2CN$	60	48
3	4a	C(O)Me	C(O)Me	Н	15	52
4 ^b	4b	CN	CN	Н	60	66
5	4c	CO(O)Me	CO(O)Me	Н	10	55
6	4d	C(O)Ph	C(O)Ph	Н	30	49
7	4e1 (E-isomer)	C(O)Ph	C(O)Me	Н	15	27
8	4e2 (Z-isomer)	C(O)Me	C(O)Ph			40
9 ^c	4f1 (<i>E</i> + <i>Z</i>) 1.7:1 4f2 (<i>E</i> + <i>Z</i>) 1.7:1	CO(O)Me C(O)Me	C(O)Me CO(O)Me	Н	20	61

^a For procedures see Ref. 12a.

^b Literature data—Refs. 8d,13.

^c The mixture of *E*- and *Z*-isomers.



Figure 1. Relative stability of radicals A and B.

spectra of both isomers. Thus, irradiation of the methyl protons of **4e2** (*Z*-isomer) at $\delta_{\rm H}$ = 2.64 results in a significant NOE of H4 at $\delta_{\rm H}$ = 8.15; whereas irradiation of the methyl protons of **4e1** (*E*-isomer) at $\delta_{\rm H}$ = 2.50 results in no NOE of H4 at $\delta_{\rm H}$ = 7.65. In this isomer, the phenyl group is shielding H4 (Fig. 2). Similar NOE experiments were carried out on the mixture of **4f**: irradiation of the H-4 proton at $\delta_{\rm H}$ = 8.21 causes NOE on the protons of CO(O)CH₃ at $\delta_{\rm H}$ = 3.90, while irradiation of the H-4 proton at $\delta_{\rm H}$ = 7.93 results in an NOE of the protons of C(O)CH₃ $\delta_{\rm H}$ = 2.48. The latter result reveals that carbomethoxy deshields H4 more than acetyl does (Fig. 2).

As additional structural proof, we synthesized **4c** independently from isatin and dimethyl malonate using the Knoevenagel method reported by Jones.^{8a}

Unfortunately, attempts to increase the yields by employing the co-oxidant $Cu(OAc)_2$ were unsuccessful and the yields of **4** were virtually unchanged.^{12b} Copper acetate in conjunction with $Mn(OAc)_3$ is known to increase the rate of oxidation of the intermediate secondary radical (i.e., **A**).^{1a,15}



Figure 2. Structures of E- and Z-isomers of 4e and 4f.

In summary, we have described the Mn(III)-promoted free radical addition of active methylene compounds to 2-nitroindole followed by a spontaneous in situ Nef reaction to provide a novel synthesis of 2-oxoindolin-3-ylidenes, which have found recent utility in the synthesis of the maremycins,¹⁶ spirocyclic 2-oxindoles,¹⁷ new Cdc25 phosphatase inhibitors,¹⁸ and β -carbolines.¹⁹ In the case of active methine compounds the radical addition reaction affords the 2-nitro-3-substituted indole.

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- (a) General procedure for the synthesis of 3-alkyl-2-nitroindoles 2a-b and 2-12 oxoindolin-3-ylidenes **4a**-**f**. 2-Nitroindole (1) (0.05 g; 0.309 mmol), Mn(OAc)₃·2H₂O (5 equiv, 0.414 g, 1.543 mmol), and the corresponding methylene or methylidene compound (5 equiv, 1.543 mmol) and AcOH (10 mL) were charged into a 50 mL round bottom flask. The mixture was refluxed for 10-60 min. After completion of the reaction, water was added (200 mL) and the water phase was extracted with AcOEt (3×50 mL). The organic phase was washed with water $(3 \times 100 \text{ mL})$, the solvent was removed under reduced pressure, and the residue was purified via column chromatography (SiO₂, EtOAc/hexare (1:1) or (1:2)) to afford the corresponding 3-alkyl-2-nitroindole **2a–b** or 2-oxoindolin-3-ylidene **4a–f**. (for yields see Table 1. (b) *General procedure using Cu(OAc)*₂ H_2O : 2-(2-Oxo-1.2-dihydroindol-3-ylidene)malonic acid dimethyl ester 4c (Procedure B): 2-Nitroindole (1) (0.05 g; 0.309 mmol), Mn(OAC)₃·2H₂O (5 equiv, 0.41 g, 1.543 mmol), Cu(OAC)₂·H₂O (0.1 equiv, 0.006 g, 0.0309 mmol), NaOAc·3H₂O (10 equiv, 0.42 g, 1.543 mmol), dimethyl malonate (5 equiv, 0.2 g, 1.543 mmol), and AcOH (10 mL) were charged into a 50 mL round bottom flask. The mixture was stirred and heated at 80 °C for 5 h. After completion of the reaction, water was added (200 mL) and the water phase was extracted with AcOEt (3 \times 50 mL). The organic phase was washed with water (3 \times 100 mL), the solvent was removed under reduced pressure, and the residue was purified via column chromatography (SiO2, EtOAc/hexane (1:2)) to afford 38 mg of compound 4c (47%) as an orange solid.

2-Methyl-2-(2-nitro-1H-indol-3-yl)malonic acid dimethyl ester **2a**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:2)) to give a yellow solid, 50 mg (53%), mp 120–122 °C. ¹H NMR (300 MHz, CDCI₃) δ 2.00 (s, 3H), 3.81 (s, 6H), 7.16–7.22 (m, 1H), 7.38–7.44 (m, 3H), 9.48 (br s, 1H); ¹³C NMR (75 MHz, CDCI₃) δ 22.0, 53.7, 55.4, 113.1, 117.1, 121.9, 122.9, 124.6, 128.7, 133.7, 170.5; *m*/z (El⁺ mode) 306 (M⁺, 43%), 260 (100), 247 (20), 219 (8), 201 (40), 186 (69), 157 (17), 146 (20), 115 (27), 101 (10), 84 (16), 59 (33); HRMS (El⁺ mode) calcd for C₁₄H₁₄N₂O₆M⁺ 306.0852, found 306.0852.

2-(2-Cyanoethyl)-2-(2-nitro-1H-indol-3-yl)malonic acid diethyl ester **2b**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:2)) to give a yellow solid, 55 mg (48%), mp 161-162 °C. ¹H NMR (300 MHz, CDCl₃) δ 1.22 (t, GH), 2.47 (t, 2H), 2.93 (t, 2H), 4.22 (m, 2H), 4.35 (m, 2H), 7.20 (m, 1H), 7.34-7.46 (m, 3H); ¹³C NMR (CDCl₃) δ 14.1, 30.4, 58.2, 63.2, 111.9, 113.6, 119.4, 121.3, 123.2, 125.5, 128.6, 133.6, 138.7, 168.7 m/z (EI⁺ mode) 373 (M⁺, 15%), 327 (32), 299 (6), 271 (5), 254 (12), 225 (100), 199 (21), 169 (11), 156 (9), 132 (12), 105 (16), 77 (15); HRMS (EI⁺ mode) calcd for C₁₈H₁₉N₃O₆ M⁺ 373.1274, found 373.1274.

3-(2-0xo-1,2-dihydroindol-3-ylidene)-2,4-dione **4a**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:2)) to give an orange solid, 37 mg (52%), mp 125-126 °C. ¹H NMR (300 MHz, CDCl₃) δ 2.51 (s, 3H), 2.56 (s, 3H), 6.87 (d, 1H), 6.98 (t, 1H), 7.30 (t, 1H), 7.41 (d, 1H), 8.59 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 30.3, 31.3, 77.3, 111.0, 919.8, 123.3, 125.3, 126.8, 132.4, 142.9, 149.0, 168.6, 198.8, 201.5; *m/z* (El* mode) 229 (M*, 76%), 214 (21), 187 (28), 172 (100), 159 (27), 144 (44), 130 (18), 116 (29), 89 (17); HRMS (El* mode) calcd for C₁₃H₁₁NO₃ M* 229.0739, found 229.0737.

2-(2-Oxo-1,2-dihydroindol-3-ylidene)malonitrile **4b**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:1)) to give a red solid, 40 mg (66%), mp 228-229 °C (lit.^{8d} mp 235-238 °C). ¹H NMR (300 MHz, DMSO-d₆) δ 6.91 (d, 1H), 7.11 (t, 1H), 7.55 (t, 1H), 7.85 (d, 1H), 11.2 (s, 1H); ¹³C NMR (75 MHz, DMSO-d₆) δ 80.6, 111.56, 111.64, 113.1, 118.8, 122.9, 125.9, 137.8, 146.5, 150.7, 163.8.

2-(2-0xo-1,2-dihydroindol-3-ylidene)malonic acid dimethyl ester **4c**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:1)) to give an orange solid, 44 mg (55%), mp 141–142 °C. ¹H NMR (300 MHz, CDCl₃) δ 3.92 (s, 3H), 3.95 (s, 3H), 6.82 (d, 1H), 7.04 (t, 1H), 7.34 (t, 1H), 7.93 (br s, 1H), 8.39 (d, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 53.4, 53.5, 110.4, 119.9, 123.3, 129.0, 129.5, 133.7, 135.4, 143.6, 163.5, 166.3, 167.5; *m*/z (El⁺ mode) 261 (M⁺, 100%), 230 (70), 202 (23), 170 (46), 162 (17), 143 (27), 130 (21), 115 (24), 88 (15); HRMS (El⁺ mode) calcd for C₁₃H₁₁NO₅ M⁺ 261.0637, found 261.0639.

2-(2-Oxo-1,2-dihydroindol-3-ylidene)-1,3-diphenylpropane-1,3-dione **4d**: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:2)) to give an orange amorphous solid, 53 mg (49%), mp 93–95 °C. ¹H NMR (300 MHz, CDCl₃) δ 6.72 (d, 1H), 6.79 (t, 1H), 6.95 (d, 1H), 7.19 (t, 1H), 7.47–7.67 (m, 5H), 8.09 (d, 1H), 8.18 (d, 2H), 8.28 (d, 2H), 8.77 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 111.1, 119.8, 122.9, 124.7, 128.7, 128.93, 128.97, 129.4, 130.1, 130.3, 131.1, 131.6, 133.8, 134.3, 134.4, 135.4, 135.6, 142.6, 147.9, 167.8, 192.1, 192.2; m/z

(El⁺ mode) 353 (M⁺, 19%), 296 (4), 248 (7), 220 (6), 122 (45), 105 (100), 84 (86); HRMS (El⁺ mode) calcd for $C_{23}H_{15}NO_3$ M⁺ 353.1052, found 353.1053. (E)-2-(2-Oxo-1,2-dihydroindol-3-ylidene)-1-phenylbutane-1,3-dione 4e1: This was purified via column chromatography (SiO2, CHCl3/hexane (1:2)) to give an orange amorphous solid, 24 mg (27%). ¹H NMR (300 MHz, CDCl₃) δ 2.50 (s, 3H), 6.78 (d, 1H), 7.03 (t, 1H), 7.31 (t, 1H), 7.44-7.52 (m, 2H), 7.56-7.68 (m, 2H), 8.00-8.12 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 30.2, 110.8, 119.9, 123.1, 125.6, 128.7, 129.16, 129.23, 130.4, 132.5, 133.8, 134.3, 136.1, 143.1, 147.4, 168.0, 193.9, 198.6; HRMS (EI⁺ mode) calcd for C₁₈H₁₃NO₃ M⁺ 291.0896, found 291.0894. (Z)-2-(2-Oxo-1,2-dihydroindol-3-ylidene)-1-phenylbutane-1,3-dione 4e2: This was purified via column chromatography (SiO₂, CHCl₃/hexane (1:2)) to give an orange amorphous solid, 36 mg (40%). ¹H NMR (300 MHz, CDCl₃) δ 2.64 (s, 3H), 6.78 (m, 2H), 6.86 (d, 1H), 7.18 (t, 1H), 7.44–7.52 (m, 3H), 7.62–7.70 (m, 1H), 8.10–8.20 (m, 2H), 8.71 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 30.8, 111.0, 119.9, 123.0, 124.8, 128.7, 129.6, 130.4, 130.7, 131.5, 133.9, 134.3, 135.5, 142.4, 149.4, 168.3, 192.0, 199.9; *m*/*z* (El⁺ mode) 291 (M⁺, 11%), 220 (6), 122 (33), 105 (58), 84 (100); HRMS (EI⁺ mode) calcd for C₁₈H₁₃NO₃ M⁺ 291.0896, found 291.0894. (E)- and (Z)-3-Oxo-2-(2-Oxo-1,2-dihydroindol-3-ylidene)butyric acid methyl ester (mixture E:Z-1.7:1) 4f: This was purified via column chromatography (SiO₂, EtOAc/hexane (1:1)) to give an orange amorphous solid, 46 mg (61%). Z-isomer (4f1): ¹H NMR (300 MHz, CDCl₃) & 2.48 (s, 3H), 3.96 (s, 3H), 6.82 (d, 1H), 6.98 (t, 1H), 7.32 (t, 1H), 7.93 (d, 1H), 8.49 (br s, 1H). E-isomer (4f2): ¹H NMR (300 MHz, CDCl3) & 2.56 (s, 3H), 3.90 (s, 3H), 6.86 (d, 1H), 7.03 (t, 1H), 7.32 (t, 1H), 8.21 (d, 1H), 8.57 (br s, 1H).

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